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How did chemistry become biology? Or: can we make life in the lab?

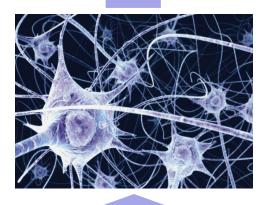
Sijbren Otto

Complexity → **Emergent** behaviour





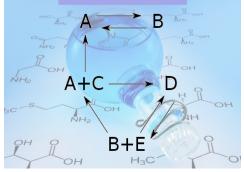


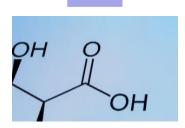






Systems chemistry







THE BIG QUESTIONS IN SCIENCE



How did life start?

Can we make life?

Can we achieve Darwinian evolution in a fully synthetic chemical system?





What is life?



NASA definition:

life is a self-sustained chemical system capable of undergoing Darwinian evolution

But...



Lonesome George



Mule



Characteristics of Life

metabolism

Compartmentalised

Dissipates energy

• Stable yet out of equilibrium -

- Ability to <u>reproduce</u>
- Errors in reproduction followed by Darwinian selection -
- Ability to adapt

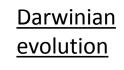


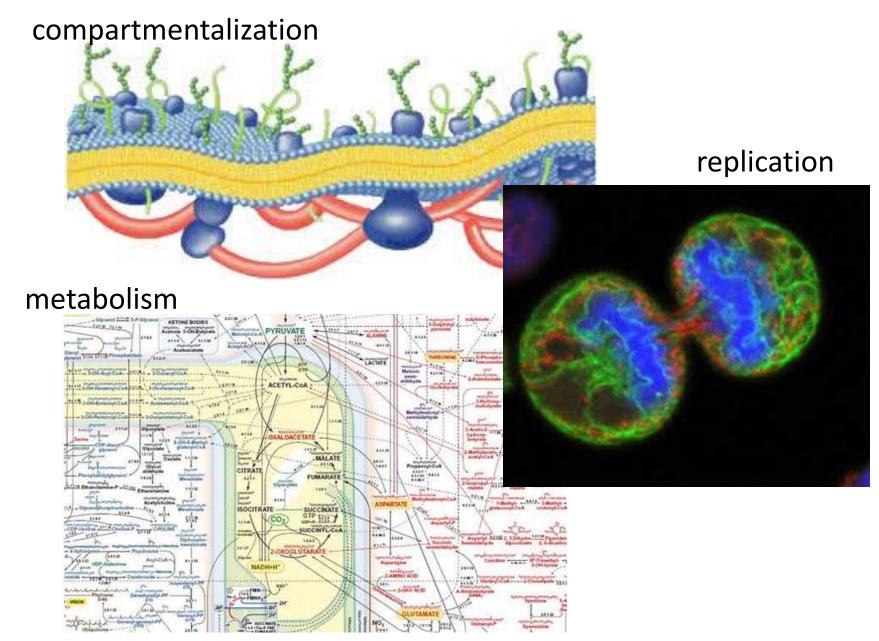
Would we recognise a chemical system that has these characteristics as being alive?

Is there a clear separation between life and non-life or are these extremes on a continuum?











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Searching for clues to the origin of life

top-down

what can we learn from current life?



bottom-up

what can we deduce from conditions on early earth? what can we learn from chemistry?

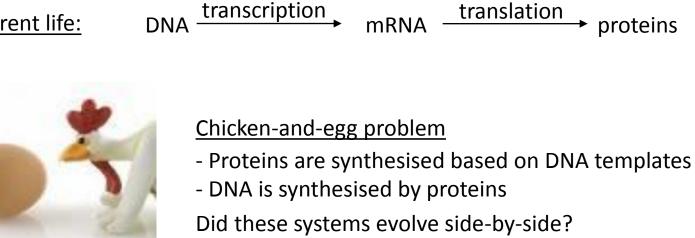




Top-down: chemical fossils in life

Assumption: current proteins merely speed up reactions that also occur spontaneously

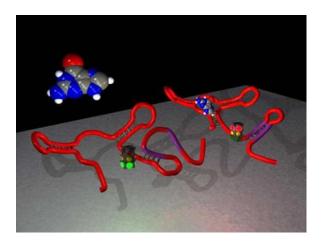
Current life:



The RNA world hypothesis:

- RNA can carry hereditary information
- RNA has a limited ability to replicate itself
- RNA can catalyse reactions

Problem: RNA is often a lousy catalyst





Bottom-up approach: prebiotic chemistry

History of the earth

Formation of the earth: 4.6 Ga (= 4.6×10^9 years)

Heavy meteorite bombardment until 3.9-3.8 Ga

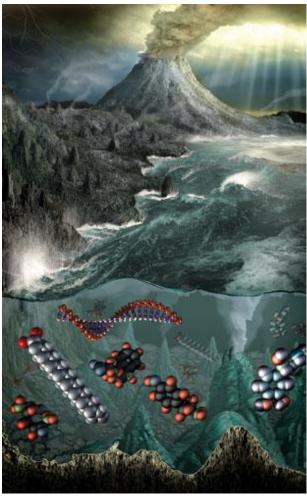
Earliest fossils: 3.8 Ga

 \rightarrow Life appeared relatively quickly

What were the ingredients of the prebiotic soup?

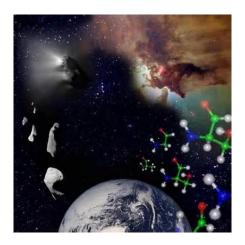
Geochemistry: what molecules occurred on early earth?

Prebiotic chemistry: aims to establish by experiment the chemical structures, reactions and pathways that may have been involved in producing the last common ancestor.





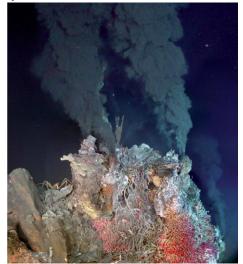
Searching for clues: Geochemistry



Probable requirement: concentrated solution of organic molecules

Where is the primordial soup most tasty for life?

- Oceans, but perhaps too dilute (so maybe in coacervates)
- Hot start: hydrothermal vents¹
 The most primitive forms of current life are found in hot water, but...
- Cold start: inclusions in ice
- Muddy start: absorption to clays
 - heterogeneous catalysis by metal ions
 - concentration of reactants by absorption



¹ Martin: *Nature Rev. Microbiol.* **2008**, *6*, 805

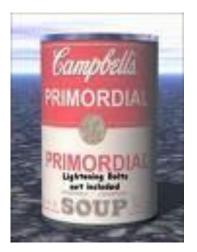
Prebiotic chemistry: Synthesis of α -amino acids

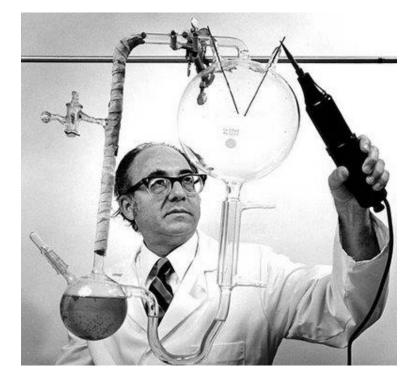
Urey-Miller experiments (1950s)

 CH_4 , NH_3 , H_2 , H_2O + electric discharge \rightarrow glycine + alanine + aspartic acid (racemic)

Flow of energy and material

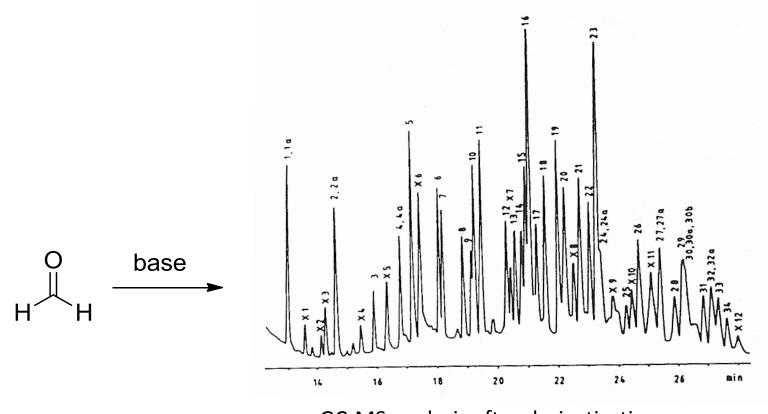
Many, many other products...







Prebiotic synthesis of sugars



GC-MS analysis after derivatisation



J. Chromatogr. **1982**, 244, 281.

Prebiotic synthesis

Nature's building blocks were around

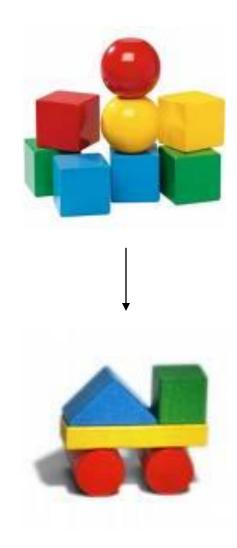
 α -amino acids

nucleobases

sugars

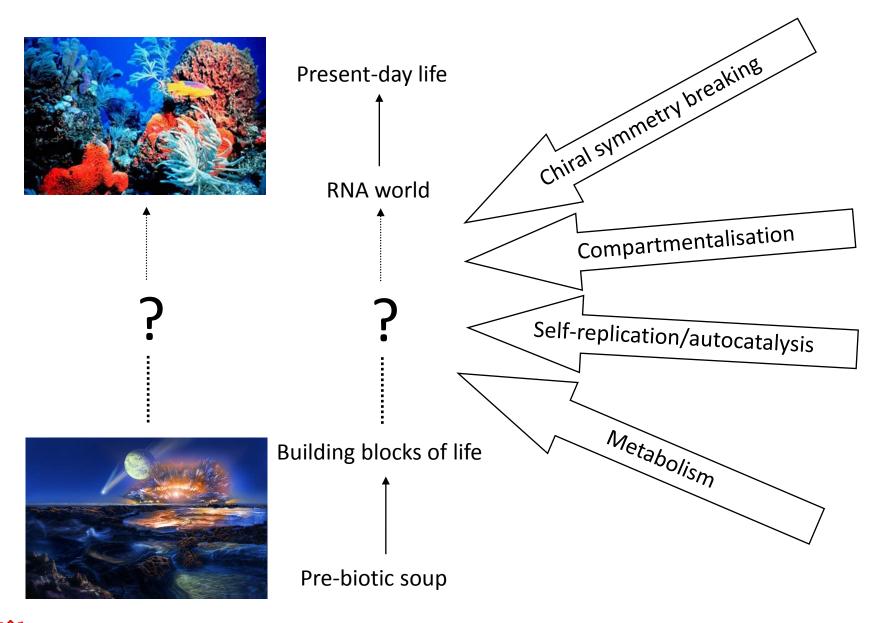
among many many other molecules!!

How were Nature's macromolecules made?





Origin of life – state of the art



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Chemistry tends to diverge

 $A + B + C \longrightarrow A-A + A-B + A-C + B-B + B-C + C-C + A-A-A + A-A-B + A-A-C + A-B-B + A-B-C + A-C-C + B-B-B +$ combinatorial explosion!

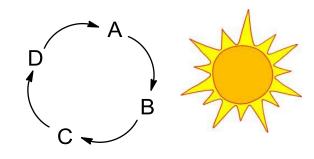
What tamed chemistry?



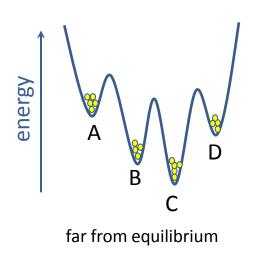
Earth before life started...

Prebiotic soup

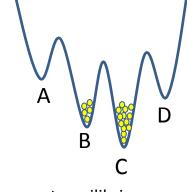
building blocks of life presentfar from equilibrium







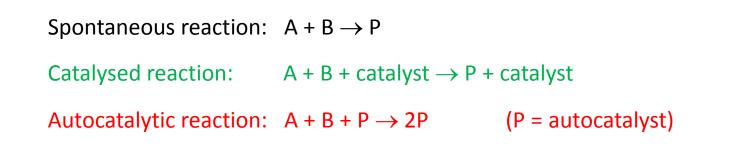


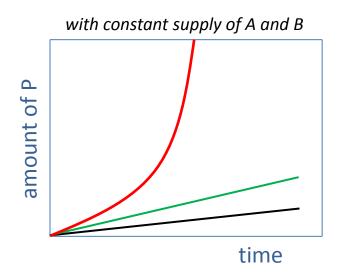


at equilibrium



What reaction is most efficient at converting molecules?



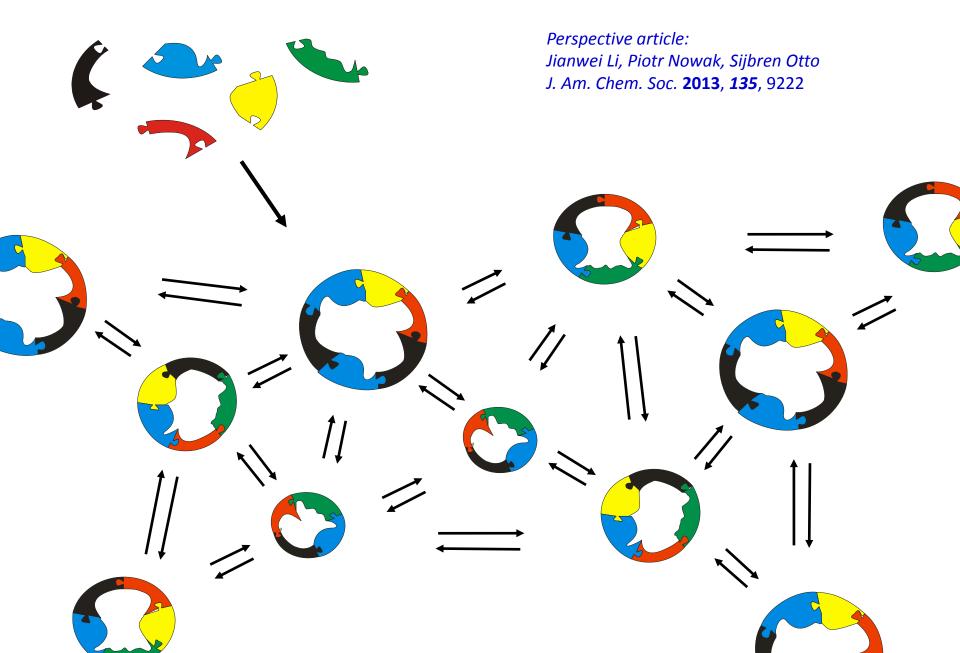


A far-from-equilibrium recycling system is likely to be dominated by products autocatalytic reactions

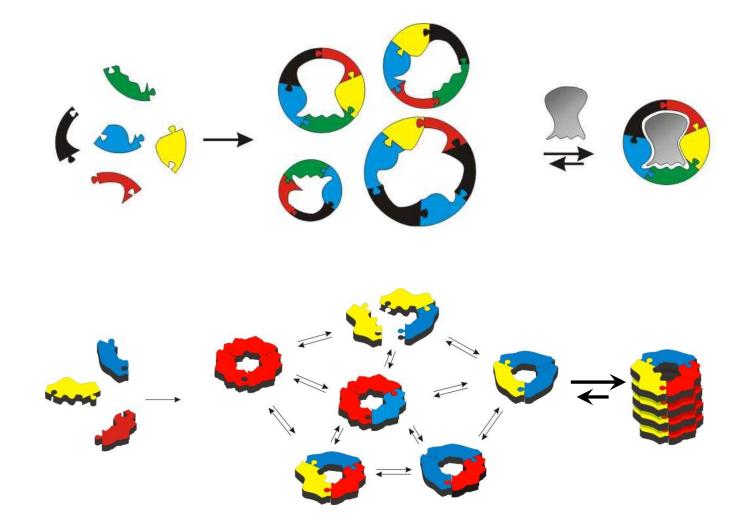
Autocatalytic systems attract matter



Dynamic Molecular Networks

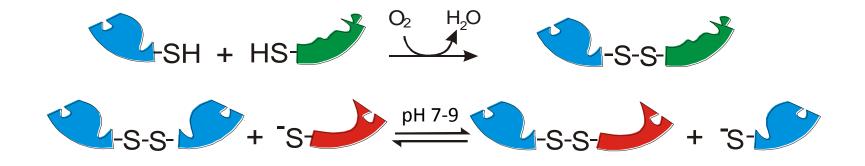


From diversity to specificity through molecular recognition





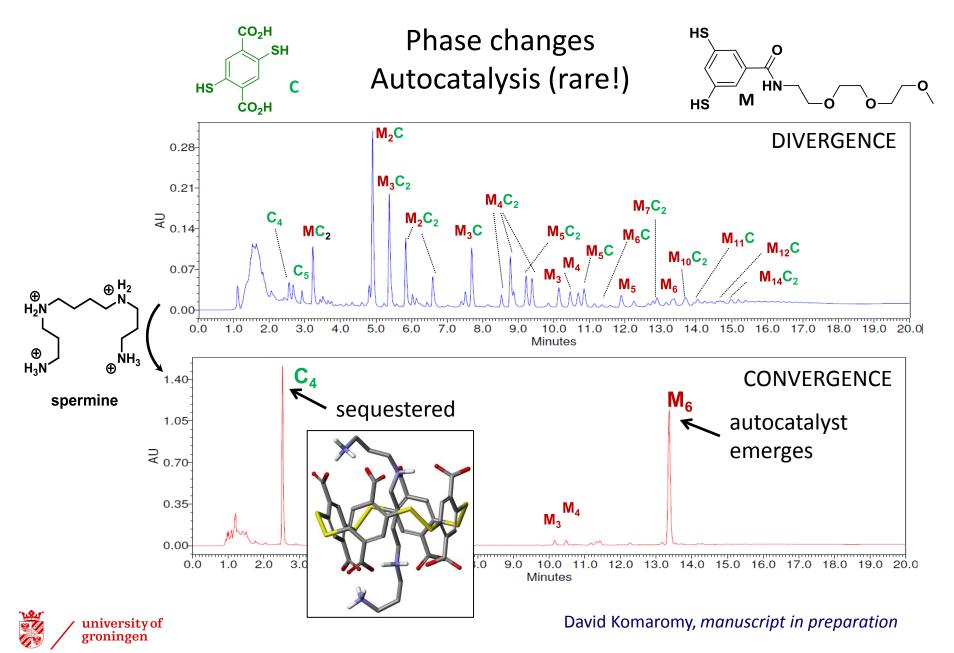
Disulfide formation and exchange





J. Am. Chem. Soc. 2000, 122, 12063

What can tame chemistry?



Can we make life in the lab?





Our recipe for life



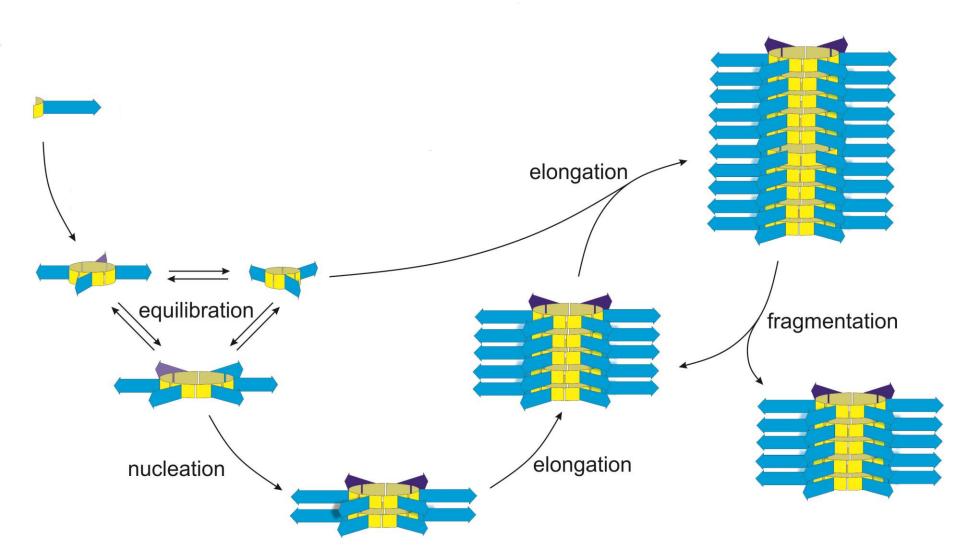
- 1 Develop an exponential replicator
- 2 Enable mutations
- 3 Operate system far from equilibrium
- 4 Allow for Darwinian evolution of the replicators
- 5 Facilitate invention of new traits (open-ended evolution)



Emergence of an exponential replicator from a soup

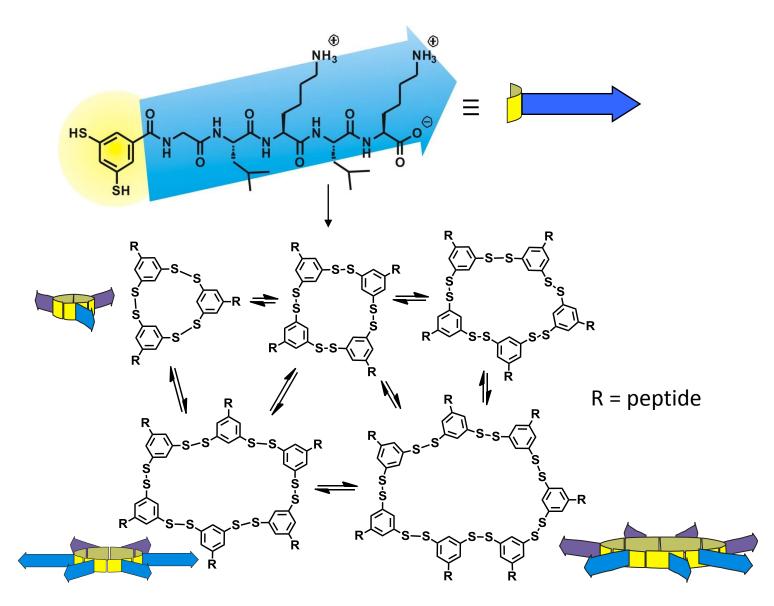
Animation: see www.otto-lab.com

Exponential replication by elongation/fragmentation



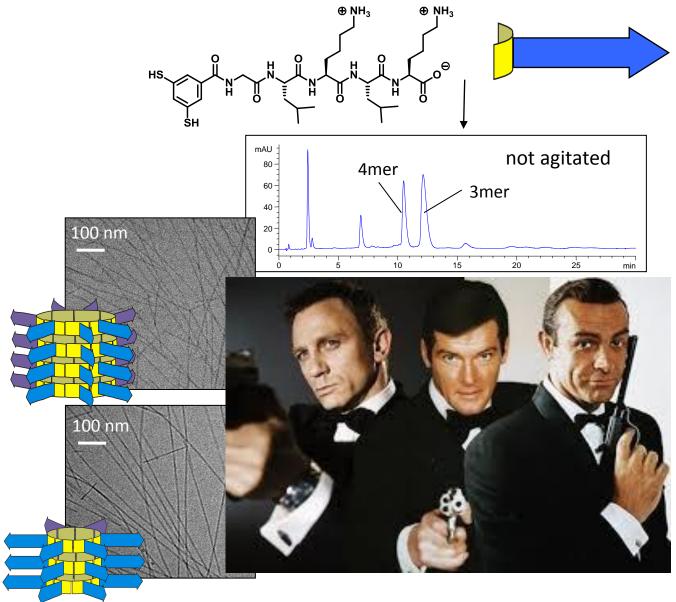


The chemical details



J. M. A. Carnall, C. A. Waudby, A. M. Belenguer, M. C. A. Stuart J. J.-P. Peyralans, S. Otto, Science 2010, 327, 1502

Mechanical energy determines replicator structure



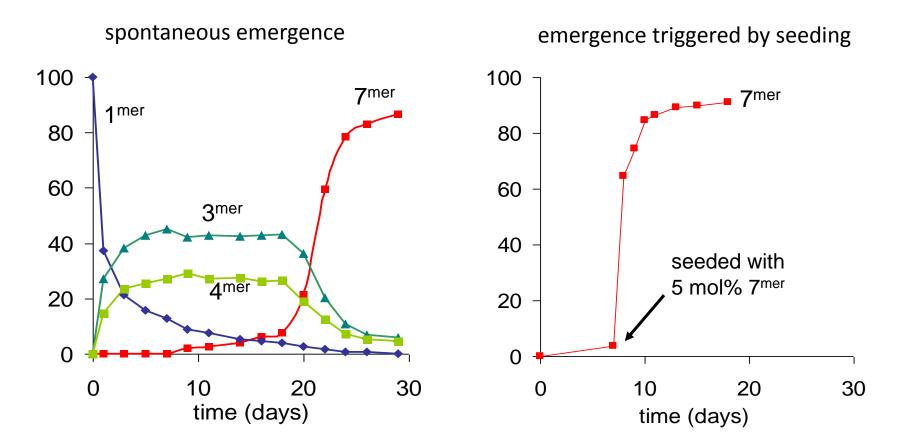


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Autocatalysis!



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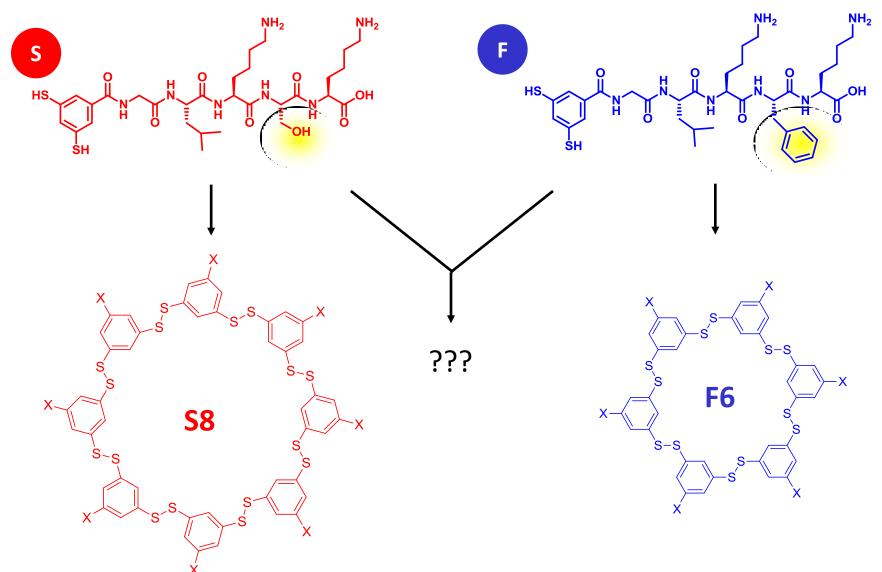
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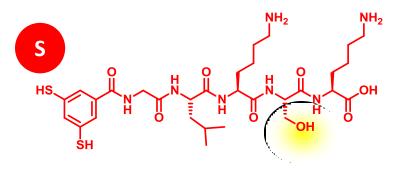
Mixing building blocks: enabling mutations

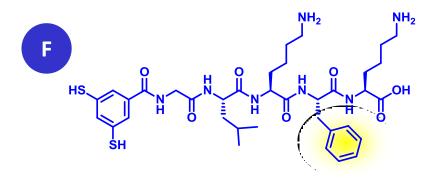


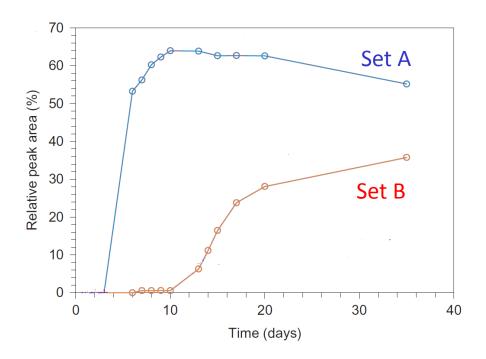


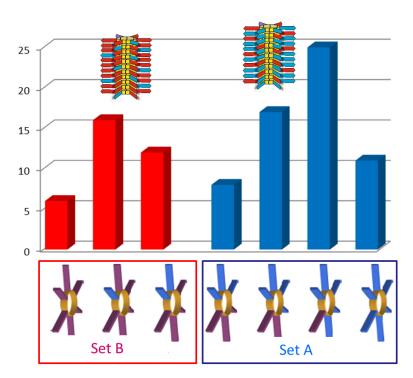
Two sets of 6mer replicators emerge sequentially

+



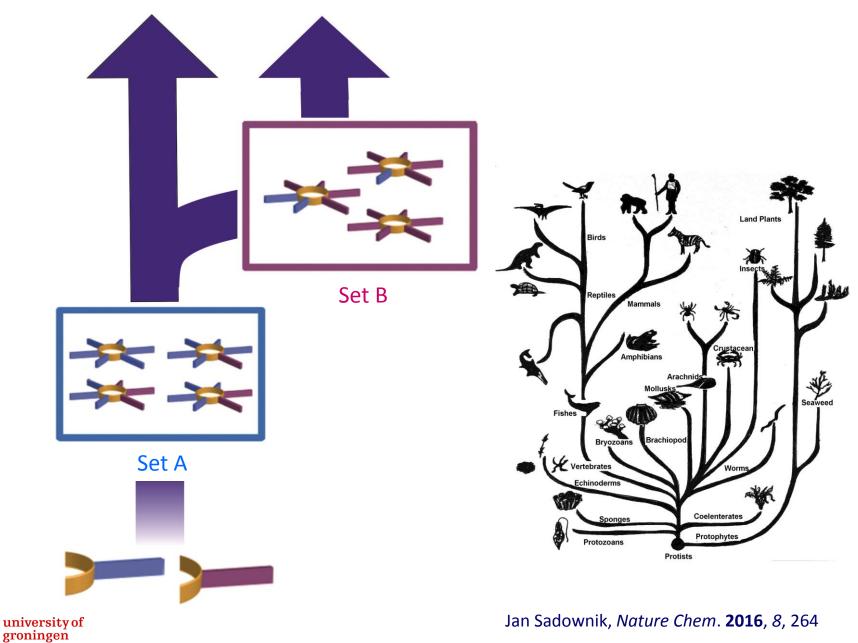








Ancestral relationship!

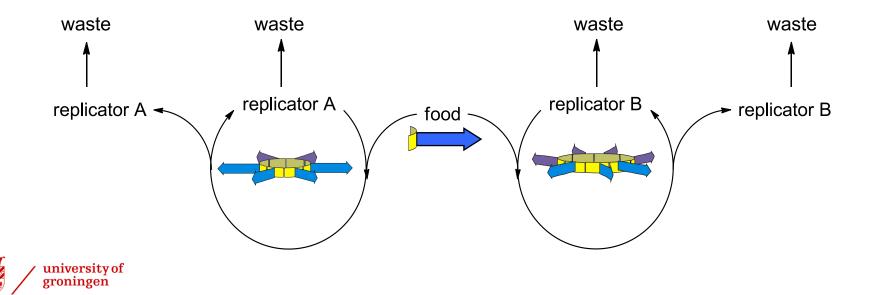




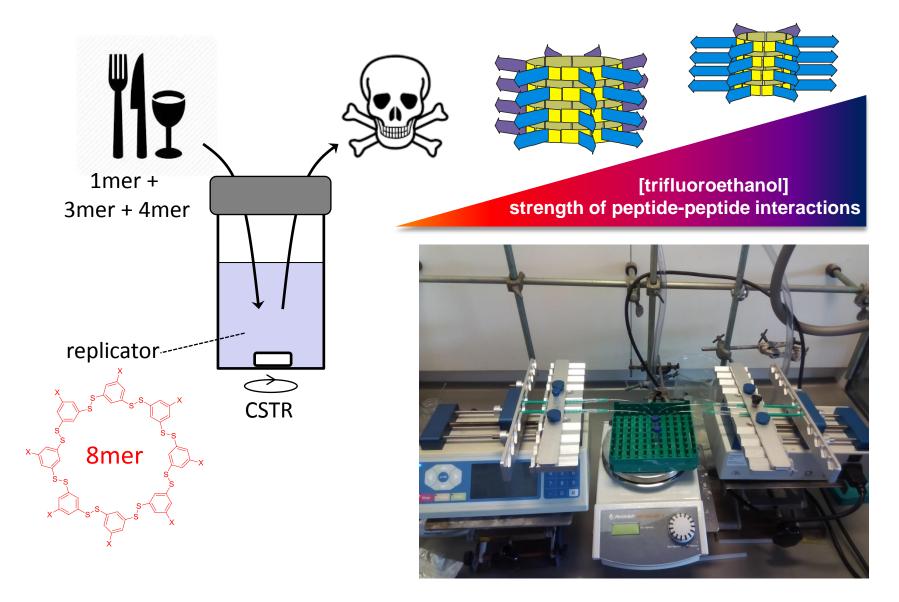
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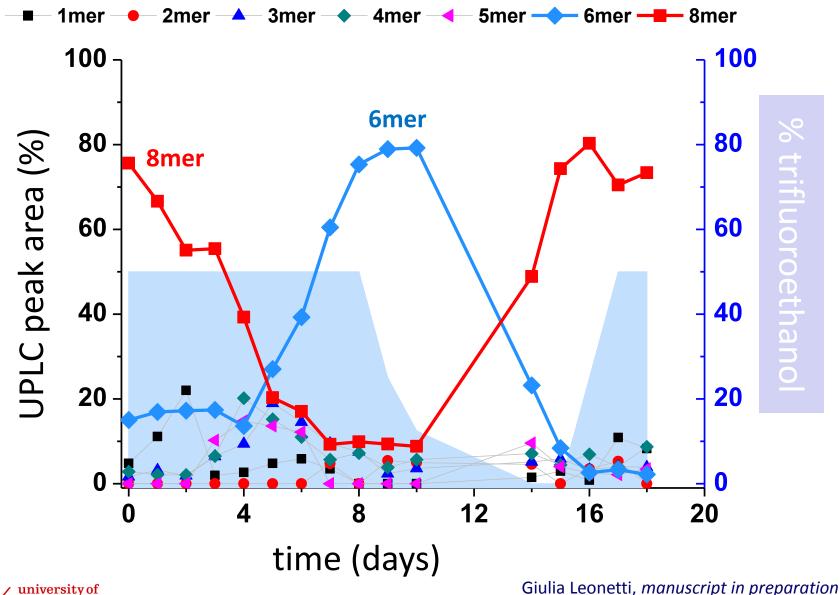


A replication-destruction regime





Adaptation to a changing environment

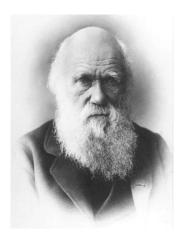


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Current open questions

Open-ended Darwinian evolution requires more than replication, variation, and selection

- Structural space must be larger than sampled space
- How to facilitate emergence of new function / step-changes in efficiency of replication??





Acknowledgements



Functional Molecular Systems Gravitation Program - The Netherlands

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Marc Stuart Jasper van der Gucht / Duc Nguyen Siewert-Jan Marrink Wim Hordijk