

Development and upscale of DME production via CO₂ direct hydrogenation

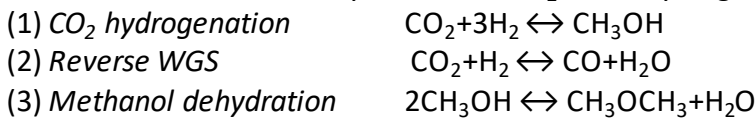
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Introduction

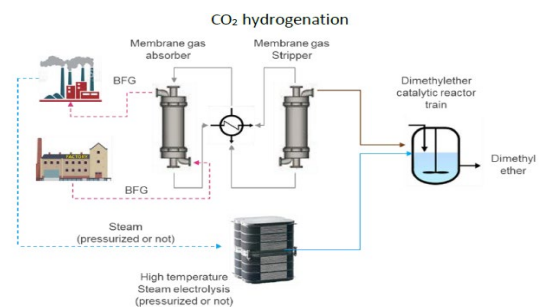
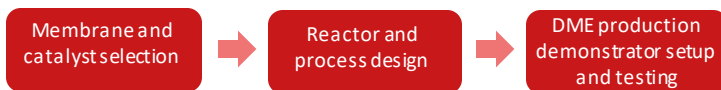
In order to reduce the CO₂ emissions, CO₂ can be captured from the industrial off-gases and converted in fuel or energy carriers. In this scenario dimethyl ether (DME) plays a key role because it is a clean burning fuel that can be a substituent of diesel fuel in diesel engines. However DME synthesis through the CO₂ direct hydrogenation route seems to be not competitive with respect to the traditional feedstocks due to thermodynamics limitations: CO₂ favors water formation through reverse WGS reaction, hindering DME production and causing catalyst deactivation.

Project summary

Three main reactions are involved in DME synthesis via CO₂ direct hydrogenation:



The development of a membrane reactor for H₂O in situ removal may bring beneficial effect on CO₂ conversion and DME yield. Alumina-carbon molecular sieve membranes seem to be the most promising, due to their hydrophilicity and pore size. Membranes will be prepared and characterized in order to find the preparation conditions that lead to the optimal properties in terms of water permeation and selectivity. Then, a bifunctional catalyst like Cu-ZnO-Al₂O₃/HZSM-5 will be selected and optimized. After a proper choose of the catalyst and the membranes, the reactor and the overall process can be designed. In the end, a DME experimental setup will be constructed and tested.



Project goals

- Develop hydrophilic CMMS allowing water removal from the reaction zone
- Demonstrate a multi-tubular membrane reactor at TRL5 up to 40 bar and 250 °C
- Increase the yield of DME by 28% per pass compared with conventional systems working at the same conditions (yield 15% per pass)

Contact information

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