

## Preparation of Dispersed Oxide by Flame Spray Pyrolysis Method for CO<sub>2</sub> Hydrogenation to methanol

Carbon dioxide (CO<sub>2</sub>) as the one of the greenhouse gases has a negative impact on climate change and ocean acidification. The chemical transformation of CO<sub>2</sub> to high calorific fuels is particularly attractive. A series of metal (Cu, Pt, Pd, Ni) supported on oxide surface has been detected its catalytic performance in CO<sub>2</sub> hydrogenation. The interface between metal and dispersed oxide and the size of active phase are crucial to control methanol selectivity. It is reported that Pd supported on TiO<sub>2</sub> displayed the highest activity for CO<sub>2</sub> hydrogenation among Pd supported on Al<sub>2</sub>O<sub>3</sub>, SiO<sub>2</sub>, TiO<sub>2</sub>, and MgO. The selectivity of CO<sub>2</sub> hydrogenation on Pd/oxides catalysts was shown to largely depend on Pd particle size. CO and CH<sub>3</sub>OH were produced on poorly dispersed Pd, while CH<sub>4</sub> was the main product on highly dispersed Pd<sup>[1]</sup>. While in Pd/In<sub>2</sub>O<sub>3</sub> system, a controlled co-precipitation method has been developed to replace the indium atoms in the active In<sub>3</sub>O<sub>5</sub> with the palladium atoms, which facilitates H<sub>2</sub> activation and further to enhance the methanol product selectivity<sup>[2]</sup>. Thus, how to construct the active interface and efficient particle size is a challenge.

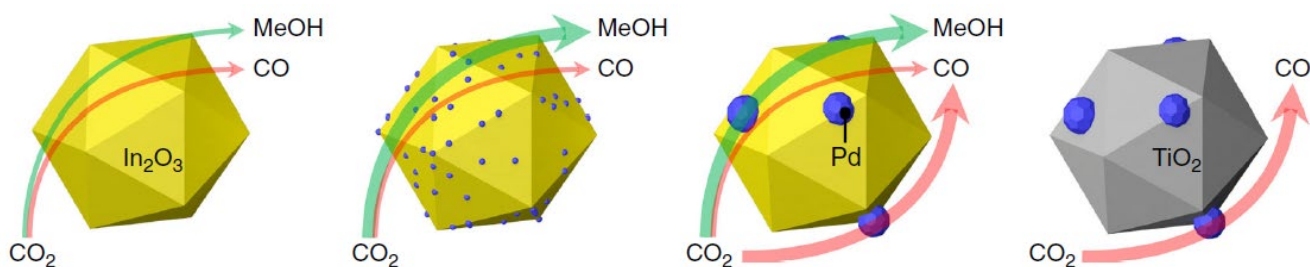


Figure 1. Schematic representation of the distinct role of palladium in equilibrated CP and DI systems with In<sub>2</sub>O<sub>3</sub> and Pd-TiO<sub>2</sub> as references<sup>[2]</sup>

In this project, we would like to work in the direction of flame spray pyrolysis as a technique to develop highly dispersed oxides with transition metal (like Fe, Co, Ni, Cu, Pd, Pt) doped into them as model systems for CO<sub>2</sub> hydrogenation. All kinds of characterizations like FTIR, Raman, TPR, XPS (including NAP XPS) and EXAFS/XANES may be required, so as to detect the size effect of oxides on CO<sub>2</sub> hydrogenation reaction. The size effect and the interface function will be able to investigate in detail the reactions occurring at the catalytic surface, contributing to the development of better catalytic processes for renewable energy storage.

### Possible projects

1. Develop and understand the size effect of active sites prepared by flame spray pyrolysis method for CO<sub>2</sub> hydrogenation to methanol.
2. Construct the active interface between transition metal and dispersed oxide to improve the catalytic performance.

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### References:

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[2] Frei M S, Mondelli C, García-Muelas R, et al. *Nat Commun* **2019**, 10(1): 1-11